

## DROP TEST CHLORIDE (1 drop = 10 ppm)

Instr. #5067O

### COMPONENTS:

- 1 x 5067O Instruction
- 1 x 9198O Sample Tube, Graduated (25 mL) w/ cap & orange dot, plastic
- 1 x R-0630-C Chromate Indicator, 2 oz, DB
- 1 x R-0638O-A Phenolphthalein Indicator, .75 oz w/ orange cap, DB
- 1 x R-0687O-C Sulfuric Acid .12N, 2 oz w/ orange cap, DB
- 1 x R-0706-C Silver Nitrate Reagent, 2 oz, DB

**TO ORDER REPLACEMENT PARTS AND REAGENTS CALL TOLL-FREE  
800-TEST KIT (800-837-8548).**

### PROCEDURE:

**CAREFULLY READ AND FOLLOW PRECAUTIONS ON REAGENT LABELS.  
KEEP REAGENTS AWAY FROM CHILDREN.**

NOTE: When dispensing reagents from dropper bottles, **always** hold bottle in a vertical position.

### Chloride Test

NOTE: When sulfite content of sample water to be tested exceeds 10 ppm, the sulfite should be oxidized to prevent interference in test. A 25 mL water sample is first adjusted to the appropriate pH, then 1 mL (or 25 drops) of R-0649 Hydrogen Peroxide Solution (sold separately) is added and thoroughly mixed. Continue with the rest of the procedure.

1. Rinse and fill 25 mL sample tube (#9198O) to 25 mL mark with water to be tested.
2. Add 2 drops R-0638O Phenolphthalein Indicator. Swirl to mix. If sample turns pink (Fig. 1), add R-0687O Sulfuric Acid .12N dropwise, swirling after each drop, until color changes from pink to colorless.

3. Add 5 drops R-0630 Chromate Indicator. Swirl to mix. Sample will turn yellow (Fig. 2).

4. Add R-0706 Silver Nitrate Reagent dropwise, swirling and counting after each drop, until color changes from yellow to milky salmon (brick red) (Fig. 3).

NOTE: A white precipitate will form as R-0706 Silver Nitrate Reagent is added to the sample. Do not add enough R-0706 Silver Nitrate Reagent to give a brown color. First change from yellow to a milky salmon (brick red) is the endpoint.

5. Multiply drops of R-0706 Silver Nitrate Reagent by 10. Record as parts per million (ppm) chloride ( $\text{Cl}^-$ ).



Fig. 1



Fig. 2



Fig. 3



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